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The mark scheme gives examiners: ? an idea of the types of response expected ? how individual marks are to be awarded ? the total mark for each question ? examples of responses that should NOT receive credit. / means that the responses are alternatives and either answer should receive full credit.

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~~Mark Scheme (Results) Summer 2013 – Edexcel~~

Mark schemes are prepared by the Principal Examiner and considered, together with the relevant questions, by a panel of subject teachers. This mark scheme includes any amendments made at the standardisation events which all examiners participate in and is the scheme which was used by them in this examination. The standardisation process ensures that the mark scheme covers the candidates' responses to questions and that every examiner understands and applies it in the same correct way.

~~A-level Chemistry Mark scheme Unit 02 – Chemistry in ...~~
comes out as 9.5×10^{18} and 1 mark should be given] [give 2 marks if equation has been correctly (and usefully)

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rearranged in some form] rearranging: $= 0.0445 \times 6.022 \times 10^{23} = 1.20 \times 10^{18}$ atoms
 $22400 \times \text{number of atoms of } ^{40}\text{K} = (5.4 / 39.102) \times (0.0117 / 100) \times 6.022 \times 10^{23} = 9.7 \times 10^{18}$
 $N_{^{40}\text{Ar}} = 0.105 \times N_{^{40}\text{K}} \times (e^{-t} - 1) N_{^{40}\text{Ar}} e^{-t} = + 1$

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communication as well as the standard of the

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General Certificate of Education (A-level) Chemistry – Unit 4:

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Kinetics, Equilibria and Organic Chemistry – June 2013 3 :
Question Marking Guidance Mark Comments: 1(a)(i) 2 or two
or second or [E] 2. 1 . 1(a)(ii) 1 or one or ...

~~A-level Chemistry Mark scheme Unit 04 – Kinetics ...~~

Version 1.0. General Certificate of Education (A-level) June 2013. Chemistry (Specification 2420) CHEM5 Unit 5: Energetics, Redox and Inorganic Chemistry. Final. Mark Scheme. Mark schemes are prepared by the Principal Examiner and considered, together with the relevant questions, by a panel of subject teachers.

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Chemistry (4CH0) Paper 2CR 1 mark for each line correct
5 (b) i ii C 5H 12 ... Order Code UG035550 Summer 2013 For
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F324 Mark Scheme June 2013 3 Question Answer Marks
Guidance 1 (a) (i) propane-1,2,3-triol 1 ALLOW absence of
'e' after 'propan' ALLOW 1,2,3-propanetriol ALLOW
absence of hyphens 1, 2 and 3 must be clearly separated:
ALLOW full stops: 1.2.3 OR spaces: 1 2 3 DO NOT ALLOW
123 IGNORE glycerol (ii)~~

~~Advanced GCE Unit F324: Rings, Polymers and Analysis~~

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Chemistry A. Advanced Subsidiary GCE Unit F322: Chains, Energy and Resources. June 2013. OCR (Oxford Cambridge and RSA) is a leading UK awarding body, providing a wide range of qualifications to meet the needs of candidates of all ages and abilities. OCR qualifications include AS/A Levels, Diplomas, GCSEs, Cambridge Nationals, Cambridge Technicals, Functional Skills, Key Skills, Entry Level qualifications, NVQs and vocational qualifications in areas such as IT, business, languages, ...

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MARK SCHEME for the May/June 2013 series 0620
CHEMISTRY 0620/31 Paper 3 (Extended Theory), maximum
raw mark 80 This mark scheme is published as an aid to

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teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not

~~0620 s13 ms 31 — GCE Guide~~

F325 Mark Scheme June 2013 3 Question Answer Marks
Guidance 1 (a) (The enthalpy change that accompanies) the formation of one mole of a(n ionic) compound from its gaseous ions (under standard conditions) 2 IGNORE 'energy needed' OR 'energy required' ALLOW as alternative for compound: lattice, crystal, substance, solid Note: 1st mark requires 1 mole

~~Advanced GCE Unit F325: Equilibria ... — A Level Chemistry~~

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Course Name: Chemistry Course Code: 4CH0 ... June 2013.

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Paper 1: Question Paper 1CR Solution: Mark Scheme 1CR

Paper 2: Question Paper 2C Solution: Mark Scheme 2C

Paper 2: Question Paper 2CR

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